

61513

V Semester B.Sc. Degree Examination, April/May 2023

(CBCS – Semester Scheme)

BIOCHEMISTRY

Biochemistry Paper – V

Time : 3 Hours

Max. Marks : 70

- Instructions :**
- 1) This Paper is for the students of new syllabus 2014-15.
 - 2) The question paper has **two** parts Part – A and Part – B.
 - 3) Answer **any eight** questions from Part – A.
 - 4) Answer **any nine** questions from Part – B.

PART – A

Answer **any eight** of the following questions. **Each** question carries **2** marks : (8×2=16)

1. Define anomer with an example.
2. Write the structure of isomaltose.
3. What are glycosaminoglycans ? Give an example.
4. Mention any two biological importance of waxes.
5. What is iodine number ? Give its significance.
6. What are androgens ? Mention their functions.
7. Write any two biological importance of vasopressin.
8. What are ampholytes ? Give an example.
9. Give the relationship between standard free energy change and equilibrium constant.
10. State First law of thermodynamics.
11. What are redox reactions ? Give an example.
12. Write the structure of ATP.



P.T.O.



PART – B

Answer **any nine** of the following questions. **Each** question carries **6** marks : (9×6=54)

13. a) Elucidate the open chain structure of glucose.
b) Write the structure of Fructose –1, 6 – diphosphate. Mention their function. (4+2)
14. a) Write the partial structure of hyaluronic acid and chondroitin-4-sulphate. Mention their biological importance.
b) List out any two biological importance of carbohydrates. (4+2)
15. a) What are lipoproteins ? Mention their types and functions.
b) Give the biological significance of cholic acid. (4+2)
16. a) Explain the ABO blood grouping system.
b) What is PUFA ? Give an example. (4+2)
17. a) Enumerate the salient features of fluid mosaic model.
b) What are prostaglanins ? (4+2)
18. a) Why ATP is energy currency of the cell ?
b) Write the structure of phosphatidyl inositol. (4+2)
19. a) How does an amino acid reacts with Ninhydrin and formaldehyde ?
b) What are aromatic amino acids ? Give an example. (4+2)
20. a) Explain the forces that stabilises the tertiary structure of proteins.
b) What is meant by oxidative rancidity ? (4+2)
21. a) How are proteins classified based on composition ? Explain.
b) What are endergonic reactions ? Give an example. (4+2)
22. a) Explain α -helical structure of proteins.
b) How is sakaguchi's test carried out ? (4+2)





23. a) Explain the salient features of oxidative phosphorylation. (4+2)
b) Define P : O ratio.
24. a) What are high energy compounds ? Mention any three examples. (4+2)
b) What are cytochromes ? Give their importance.
25. a) Explain the stages of energy transformation in living systems. (4+2)
b) Mention the number of ATP molecules produced by NADH and FADH₂.



3. What is the difference between...
4. Mention the...
5. What is the...
6. What are androgens ? Mention...
7. Write any two biological...
8. What are erythrocytes ? Give...
9. Give the relationship between standard free energy change and equilibrium constant.
10. State First law of thermodynamics.
11. What are redox reactions ? Give an example.
12. Write the structure of ATP.